

Alanya Alaaddin Keykubat University | Rafet Kayış Faculty of Engineering  
 Mechanical Engineering Department  
 2022-2023 Fall Semester  
**SYLLABUS**

<b>Code/Name</b>	MEC 205 / Thermodynamics I
<b>Type</b>	Required
<b>Credit/ECTS</b>	5/5
<b>Hour per Week</b>	3 (3+0+0)
<b>Level/Year</b>	Undergraduate/2
<b>Semester</b>	Fall
<b>Classroom</b>	A 103 and A 403
<b>Content</b>	Basic concepts of thermodynamics. Properties of pure substances, ideal gases and compressibility factor. Energy, energy transfer and the first law of thermodynamics. Energy analysis for closed systems and control volumes. The second law of thermodynamics. Heat engines and refrigerators. Carnot cycle. Entropy, entropy generation, and entropy balance. Isentropic efficiencies of steady-flow devices. Exergy analysis.

**Prerequisites**

**Textbooks**

**Primary**

Y A Cengel, M A Boles, M Kanoglu, *Thermodynamics: An Engineering Approach*, 10<sup>th</sup> edition, McGraw-Hill, 2023.

**Supplementary**

M J Moran, H N Shapiro, D D Borttner, M B Bailey, *Fundamentals of Engineering Thermodynamics*, 9<sup>th</sup> edition, Wiley, 2020.

**Objectives**

- To analyze energy systems using the first law of thermodynamics
- To analyze energy systems using the second law of thermodynamics
- To assess performance of energy conversion devices

**Course Outcomes**

In this course you will be able to:  
 C01 Identify different forms of energy and energy conversion processes  
 C02 Use energy balance on closed systems to formulate and solve different thermodynamic problems  
 C03 Use mass and energy balance to solve steady-flow and unsteady flow problems  
 C04 Describe operation and principles of heat engines, steam cycle, refrigerators, and heat pumps  
 C05 Perform entropy and exergy balance on closed systems and control volumes  
 C06 Calculate isentropic efficiencies of turbines, compressors, nozzles, and pumps

**Weekly Schedule of Topics**

W	Topic
1	Basic concepts of thermodynamics, temperature
2	Energy and the first law of thermodynamics
3	Properties of pure substances, ideal gases, compressibility factor
4	Closed system energy analysis for real substances
5	Closed system energy analysis for ideal gases and incompressible substances
6	Mass balance, energy analysis for steady-flow devices
7	Energy analysis for steady-flow devices and unsteady systems
8	The second law of thermodynamics, irreversibility
9	Heat engines, refrigerators, Carnot cycle
10	Entropy, entropy transfer, entropy generation

11	Isentropic efficiencies
12	Entropy balance
13	Exergy, exergy destruction, second-law efficiency
14	Exergy balance

**Professional Contribution** Ability to understand, analyze, and improve energy systems

**Contribution to Program Outcomes\***

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11
CO1	0	0	0	0	0	0	0	0	5	0	0
CO2	5	0	0	5	0	0	0	0	3	3	0
CO3	5	0	0	5	0	0	0	0	3	3	0
CO4	4	0	0	5	0	0	0	0	3	0	0
CO5	5	0	0	5	0	0	0	0	3	3	0
CO6	5	0	0	5	0	0	0	0	3	3	0

\* Contribution Level | 0: None | 1: Very Low | 2: Low | 3: Medium | 4: High | 5: Very High

**Special Conditions** • Students work in groups for project and presentations.

**Requirements**

<b>Evaluation</b>	Midterm Exam	40%
	Quiz, Assignment	15%
	<u>Final Exam</u>	<u>45%</u>
	Total	100%

**Rubric**

**Course Policy**

1. Students are required to attend at least 70% of the theoretical courses and 80% of the courses with lab/application sessions including add-drop period. Otherwise, you will receive a grade of DZ. Health reports and other official or nonofficial excuses are not accepted.
2. Be in the class on time. Late attendance may result in grade deductions.
3. English should always be used to communicate with one another.
4. Mobile phone should be switched off and put away during the class.
5. Illegal copies of the textbooks and other illegal course materials cannot be used for the classwork and exams.

**Cheating & Plagiarism**

- Copying or letting someone to copy your work on exams, assignments, or reports is cheating.
- Cutting and pasting text, figures and tables from the web sources or any other electronic source is plagiarism.
- A consequence of academic dishonesty is to receive a grade of FF for the course.

**Instructor**

Name/Surname	Mehmet Kanoglu	Email	mehmet.kanoglu@alanya.edu.tr
Room	121	Office Hours	Tuesday: 15:30 – 16:30 Thursday: 15:30 – 16:30

Prepared by Mehmet Kanoğlu